STEREOREGULATION IN THE PREPARATION OF 1- AND 3-MONOSUBSTITUTED 1,2,3,4,6,7,12,12b-OCTAHYDROINDOLO[2,3-<u>a</u>]QUINOLIZINES.

MAURI LOUNASMAA^{*}, REIJA JOKELA, BIRGIT TIRKKONEN AND TARJA TAMMINEN

Laboratory for Organic and Bioorganic Chemistry, Technical University of Helsinki, SF-02150 Espoo, Finland

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<u>Abstract</u> - Our recently developed method is successfully applied to the preparation of 1- and 3-monosubstituted 1,2,3,4,6,7,12,12boctahydroindolo[2,3-<u>a</u>]quinolizines possessing the C(12b)H-C(1)H and C(12b)H-C(3)H relationship, respectively, <u>cis</u> or <u>trans</u> at will. Complete ¹³C NMR data are presented for the prepared compounds.

We recently developed a new synthetic method¹ which permits the preparation of 2-monosubstituted 1,2,3,4,6,7,12,12b-octahydroindolo[2,3-a]quinolizines possessing at will the C(12b)H-C(2)H <u>cis</u> or <u>trans</u> relationship.

Here we explore the applicability of the method to the preparation of 1and 3-monosubstituted 1,2,3,4,6,7,12,12b-octahydroindolo $[2,3-\underline{a}]$ quinolizines possessing the C(12b)H-C(1)H and C(12b)H-C(3)H relationship, respectively, <u>cis</u> or <u>trans</u> at will.

RESULTS AND DISCUSSION

Alkylation of 3-ethylpyridine and methyl 3-pyridylacetate with tryptophyl bromide² afforded pyridinium salts <u>1</u> and <u>2</u>, which, by NaBH₄ reduction and cyanide trapping, ³⁻⁸ were transformed to α -aminonitriles <u>3</u> and <u>4</u>, respectively. When <u>3</u> and <u>4</u> were treated with AcOH, indolo[2,3-<u>a</u>]quinolizidines <u>5</u> and <u>6</u> were obtained. A part of compounds <u>5</u> and <u>6</u> was transformed with di-<u>t</u>-butyl dicarbonate [(BOC)₂O]⁹⁻¹² to the corresponding BOC-protected indolo[2,3-<u>a</u>]-quinolizidines <u>7</u> and <u>8</u> (Scheme 1).



 $\frac{7}{8} R = CH_2CH_3$ $R = CH_2CO_2CH_3$

Scheme 1

Catalytic hydrogenation (PtO_2) of compounds <u>5</u> and <u>6</u> led to compounds <u>9</u> (60%) and <u>11</u> (5%), and <u>10</u> (75%) and <u>12</u> (15%), respectively, of which the main products (compounds <u>9</u> and <u>10</u>) possess the C(12b)H-C(3)H <u>cis</u> configuration. Compounds <u>9</u>, <u>10</u>, <u>11</u> and <u>12</u> were transformed with $(BOC)_2O$ to the corresponding BOC-protected compounds <u>13</u>, <u>14</u>, <u>15</u> and <u>16</u>, respectively. It was also verified that acid-induced cleavage (HCOOH) of the protective BOC-groups leads back to compounds <u>9</u>, <u>10</u>, <u>11</u> and <u>12</u> (Scheme 2).

By contrast, catalytic hydrogenation (PtO_2) of the BOC-protected compounds 7 and 8 led to compounds 13 (5%) and 15 (75%), and 14 (15%) and 16 (70%), respectively, of which the main products (compounds 15 and 16) possess the C(12b)H-C(3)H trans configuration. It was reverified (vide supra) that compounds 13, 14, 15 and 16 could be transformed to the corresponding non-protected compounds 9, 10, 11 and 12 by acid-induced cleavage (HCOOH) (Scheme 2).



Scheme 2

NaBH₄ reduction of pyridinium salt <u>1</u> yielded compound <u>17</u>, which by H_2O_2 oxidation and modified Polonovski reaction^{6,13-15} was transformed to compound <u>18</u>. A part of compound <u>18</u> was treated with (BOC)₂O to afford the corresponding BOC-protected compound <u>19</u> (Scheme 3).

18 led to compound 20 Catalytic hydrogenation (PtO₂) of compound possessing the C(12b)H-C(1)H cis configuration. Traces of the corresponding C(12b)H-C(1)H trans derivative (compound 23; vide infra) were detected by $(BOC)_2O$ transformed by treatment to the TLC. Compound 20 was corresponding BOC-protected compound 21, from which the non-protected compound 20 could be regenerated by acid-induced cleavage (HCOOH) (Scheme 4).

Again, in contrast to the above results, catalytic hydrogenation (PtO_2) of the BOC-protected compound <u>19</u> led to compound <u>22</u> possessing the C(12b)H-C(1)H <u>trans</u> configuration. Traces of the corresponding C(12b)H-C(1)H <u>cis</u>





19 R = CH_2CH_3

Scheme 3

derivative (compound <u>21</u>; <u>vide supra</u>) were detected by TLC. By acid-induced cleavage (HCOOH), compound <u>22</u> afforded the corresponding non-protected compound <u>23</u>. It was also verified that the $(BOC)_2O$ treatment transformed compound <u>23</u> back to the BOC-protected compound <u>22</u> (Scheme 4).

We also became interested in the behaviour of the indolo[2,3-<u>a</u>]quinolizidine skeleton possessing an exocyclic double bond at C-3 (present in many indole alkaloids) in the above described reduction conditions. For the compounds to be tested we chose (±)-deplancheine 24^{16} and its BOCprotected counterpart <u>25</u>, easily prepared from <u>24</u> by (BOC)₂O treatment (Scheme 5).

Catalytic hydrogenation (Pt0₂) of (±)-deplancheine 24 led almost possessing the C(12b)H-C(3)Hexclusively to compound 11 trans configuration, whereas the same treatment of compound 25 afforded an equimolar mixture of compounds 13 and 15 possessing the C(12b)H-C(3)H cis and trans configurations, respectively (Scheme 5). Thus, even in these



Scheme 4

cases the presence or absence of the BOC group in the indolo[2,3a]quinolizidine skeleton has its effect on the stereochemistry of the resulting products, though in the opposite way to that described above.

The ¹³C NMR data of compounds <u>4</u>, <u>6</u>, <u>8</u>, <u>10</u>, <u>12</u>, <u>14</u>, <u>16</u>, <u>17</u>, <u>18</u>, <u>19</u>, <u>21</u>, <u>22</u>, <u>24</u> and <u>25</u> are given in Fig. 1. For those of compounds <u>3</u>, <u>5</u>, <u>7</u>, <u>9</u>, <u>11</u>, <u>13</u> and <u>15</u>, see Ref. 6 (compounds <u>2</u>, <u>3</u>, <u>5</u>, <u>8</u>, <u>9</u>, <u>11</u> and <u>12</u>, respectively) and for those of compounds <u>20</u> and <u>23</u>, see Ref. 17 (compounds <u>2</u> and <u>1</u>, respectively}.

Comparison of the chemical shifts found for compounds <u>6</u>, <u>8</u>, <u>10</u>, <u>12</u>, <u>14</u>, <u>16</u>, <u>18</u>, <u>19</u>, <u>21</u>, <u>22</u>, <u>24</u> and <u>25</u> with those given earlier, 1,6 , 17 taking into account the conformational considerations relevant for indolo[2,3-<u>a</u>]quinolizidines, 1,5 provides clear evidence of the stereostructures depicted in the formulae. Drastic changes are seen in the shift values for C(60), C(7a), C(9), C(10), C(11) and C(12a) in the BOC-protected indolo[2,3-<u>a</u>]quinolizidines <u>8</u>, <u>14</u>, <u>16</u>, <u>19</u>, <u>21</u>, <u>22</u> and <u>25</u> relative to their non-protected counterparts <u>6</u>, <u>10</u>, <u>12</u>, <u>18</u>, <u>20</u>, <u>23</u> and <u>24</u>. The shift value of ~

47 ppm found for C(6) in the spectra of compounds $\underline{8}$, $\underline{16}$ and $\underline{25}$, and that of 51.3 ppm found in the spectrum of compound $\underline{14}$, are in agreement with the earlier results.⁶



Scheme 5

The clear difference between the C(6) signals (51.8 <u>vs.</u> 42.9 ppm) in the spectra of compounds <u>21</u> and <u>22</u> indicates a pronounced change in the contribution of different conformations of the indolo[2,3-<u>a</u>]quinolizidine skeleton to the conformational equilibrium of these two compounds. For a more detailed general discussion of this subject, see Refs. 1, 5 and 6.

CONCLUSIONS

The results clearly demonstrate the value of our method for the stereoselective preparation of 1- and 3-monosubstituted 1,2,3,4,6,7,12,12b-octahydroindolo[2,3-<u>a</u>]quinolizines with the C(12b)H-C(1)H and C(12b)H-C(3)H relationship, respectively, <u>cis</u> or <u>trans</u> at will.



Fig. 1

EXPERIMENTAL

IR spectra were recorded with a Perkin-Elmer 700 spectrophotometer in $CHCl_3$, if not otherwise stated. ¹H and ¹³C NMR spectra were measured with a Jeol JNM-FX 60 spectrometer working at 59.80 MHz (¹H NMR) and 15.04 MHz (¹³C NMR). The spectra were recorded in $CDCl_3$. Chemical shift data are given in ppm downfield from TMS. Abbreviations s, d, t, m, br and def are used to designate singlet, doublet, triplet, multiplet, broad and deformed, respectively. For the ¹³C NMR data of compounds 3, 5, 7, 9, 11, 13 and 15, see Ref. 6 (compounds 2, 3, 5, 8, 9, 11 and 12, respectively) and for the data of compounds 20 and 23, see Ref. 17 (compounds 2 and 1, respectively). For the ¹³C NMR data, see Fig. 1. Mass spectrometry (EIMS and HRMS) was done on a Jeol DX 303/DA 5000 instrument.

Preparation of compound 3

Hydrochloric acid (6 N, 1.5 ml) was added dropwise to a stirred solution of KCN (1.13 g, 17.4 mmol) in H₂O (1.5 ml), layered with Et₂O (9 ml), and kept at 0°C (Ar atm). Pyridinium salt 1^{17} (1.00 g, 3.02 mmol) prepared by alkylation of 3-ethylpyridine with tryptophyl bromide², was dissolved in MeOH (2.4 ml), and NaBH₄ (127 mg, 3.32 mmol) was added during 0.5 h. The mixture was stirred at rt for 3.5 h. The Et₂O layer was separated and the aqueous layer was extracted several times with Et₂O. The combined ethereal extracts were dried (Na₂SO₄) and evaporated to give compound $\frac{3}{2}$.

Analytical data were identical with those described earlier.⁶

Preparation of compound 4

Reaction of pyridinium salt $2^{11,18}$ (1.13 g, 3.02 mmol), prepared by alkylation of methyl 3-pyridylacetate with tryptophyl bromide², with KCN (1.13 g, 17.4 mmol) and NaBH₄ (127 mg, 3.32 mmol) using the procedure described for compound <u>3</u> (vide supra) afforded compound <u>4</u>.

Y. 92%. Amorphous material.

IR: 3450 (NH), 1740 (C=O).

¹H NMR: 3.63 (3H, s, $-CO_2CH_3$), 5.57 (1H, br s, H-4'), 6.89 (1H, d, J=1.8 Hz, H-2), 7.17-7.65 (4H, m, arom. H), 8.47 (1H, br s, NH). MS: 323 (M⁺), 296, 237, 193, 179, 166, 165, 151, 144 (100%), 130; exact mass: 323.1624 (calc. for $C_{19}H_{21}N_3O_2$: 323.1634).

7622

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Preparation of compound 5
Compound 3 (0.76 g, 2.72 mmol) was dissolved in 80 ml of 50% HOAc and the
solution was stirred at rt (N_2 atm) for 3 d. It was then evaporated and
           with
                    2N
                          Na<sub>2</sub>CO<sub>3</sub>. Extraction
                                                    with CH<sub>2</sub>Cl<sub>2</sub>
                                                                        and
shaken
                                                                               drying
(Na_2SO_4) yielded the crude product of 5^{6,19}, which was purified by
column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 95:5).
                   146-148<sup>0</sup>C
                               (EtOH) (lit.
                                                    Mp. 147-148^{\circ}C^{4}, 146-148^{\circ}C^{6},
Y. 45%.
            Mp.
143 - 145^{\circ}c^{20}).
Analytical data were identical with those described earlier.<sup>6</sup>
Preparation of compound 6
Compound 4 (0.88 g, 2.72 mmol) was dissolved in 90 ml of 50% HOAc and the
sclution was stirred at rt (Ar-atm) for 23 h. Usual work-up afforded
                                                                                   the
                          compound 6^{18}, which was
       product of
crude
                                                              purified
                                                                           by column
chromatography (alumina, CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 99.5:0.5).
Y. 42%. Mp. 133<sup>o</sup>C (MeOH) (lit.<sup>18</sup> amorphous material).
IR: 3450 (NH), 1730 (C=0).
<sup>1</sup>H NMR: 3.69 (3H, s, -CO_2CH_3), 5.64 (1H, br s, H-2), 6.99-7.54 (4H,
m, arom. H), 8.14 (1H, br s, NH).
MS: 296 (1)77, 263, 237, 270 (2008), 269; exact mass: 296.2536 (care. For
C<sub>18</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>: 296.1525).
Preparation of compound 7
To compound 5 (116 mg, 0.46 mmol) in 1 ml of abs. CheCle was added p-
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dimethylamino pyridine (DMAP) (6 mg, 0.1 equiv.) and di-t-butyl dicarbonate $[(BOC)_2O]$ (120 mg, 1.2 equiv.) with stirring at rt (Ar atm). After 2 h the mixture was evaporated and purified by column chromatography (silica, CH₂Cl₂-MeOH-Et₃N, 97.75:2:0.25) to afford compound <u>7</u>. Y. 90%. Viscous oil.

Analytical data were identical with those described earlier. 6

Preparation of compound $\underline{8}$

Reaction of compound <u>6</u> (166 mg, 0.56 mmol) with DMAP (7 mg, 0.1 equiv.) and $(BOC)_2O$ (220 mg, 1.8 equiv.) using the procedure described for compound <u>7</u> (<u>vide supra</u>) gave the crude product of <u>8</u>, which was purified by column chromatography (silica, CH_2Cl_2 -MeOH-Et₃N, 99:0.75:0.25). Y. 85%. Viscous oil. IR: 1725 (C=O). ¹H NMR: 1.66 (9H, s, -C(CH₃)₃), 3.69 (3H, s, -CO₂CH₃), 5.67 (1H, br s, H-2), 7.14-7.52 (3H, m, H-8, 9, 10), 8.07 (1H, m, H-11).

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396 (M<sup>+</sup>), 339, 295, 281, 214, 170 (100%), 169, 168; exact mass:
MS:
396.2058 (calc. for CooHoeNoO4: 396.2049).
Preparation of compounds 9 and 11
Hydrogenation (MeOH, PtO<sub>2</sub>, 25 h) of compound 5 (230 mg, 0.91 mmol)
followed by separation by column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH,
95:5) yielded compounds 9 and 11.
Compound 9:
Y. 60%. Viscous oil.
Analytical data were identical with those described earlier. 6,17,21,22
Compound 11:
Y. 5%.
           Mp.
                 159-161<sup>0</sup>C (EtOH) (lit.
                                                  Mp. 159-161°C<sup>6</sup>. 160-161°C<sup>22</sup>.
157^{\circ}c^{23}).
Analytical data were identical with those described earlier.6,17,21-23
Preparation of compounds 10 and 12
Hydrogenation (MeOH, PtO<sub>2</sub>, 25 h) of compound 6 (224 mg, 0.76 mmol)
followed by separation by column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH,
97:3) yielded compounds 10 and 12.
Compound 10:
Y. 75%. Mp. 180-182<sup>O</sup>C (MeOH).
Analytical data were identical with those described earlier.<sup>11</sup>
Compound 12:
Y. 15%. Mp. 164-165°C (MeOH) (lit.<sup>18</sup> Mp. 168°C).
IR: 3420 (NH), 1740 (C=O).
<sup>1</sup>H NMR: 3.68 (3H, s, -CO<sub>2</sub>CH<sub>3</sub>), 7.00-7.55 (4H, m, arom. H), 7.81 (1H,
br s. NH).
MS 298 (M<sup>+</sup>), 297 (100%), 283, 267, 225, 197, 184, 170, 169; exact mass:
298.1683 (calc. for C_{18}H_{22}N_2O_2: 298.1681).
Preparation of compound 13
Reaction of compound 9 (133 mg, 0.52 mmol) with DMAP (6 mg, 0.1 equiv.) and
(BOC)<sub>2</sub>O (204 mg, 1.8 equiv.) using the procedure described for compound 7
(vide supra) afforded the crude product of compound 13, which was purified
by column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 97:3).
Y. 90%. Viscous oil.
Analytical
              data were identical with those described
                                                                          earlier.<sup>6</sup>
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7624

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Preparation of compound 14
Reaction of compound 10 (90 mg, 0.30 mmol) with DMAP (4 mg, 0.1 equiv.) and
(BOC) 20 (120 mg, 1.8 equiv.) using the procedure described for compound 7
(vide supra) gave the crude product of compound 14, which was purified by
column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 99:1).
Y. 90%. Viscous oil.
IR: 1730 (C=O).
1_{\rm H}
             1.64 (9H, s, -C(CH_3)_3), 3.66 (3H, s, -CO_2CH_3), 7.13-7.50
     NMR:
(3H, m, H-8, 9, 10), 8.08 (1H, m, H-11).
MS: 398 (M<sup>+</sup>), 342, 341 (100%), 297, 269, 170, 169; exact mass: 398.2219
(calc. for C<sub>23</sub>H<sub>30</sub>N<sub>2</sub>O<sub>4</sub>: 398.2206).
Preparation of compound 15
Reaction of compound 11 (35 mg, 0.14 mmol) with DMAP (1.7 mg, 0.1 equiv.)
and (BOC)<sub>2</sub>O (53 mg, 1.8
                                 equiv.) using the procedure described for
compound 7 (vide supra) afforded the crude product of compound 15, which
was purified by flash chromatography (alumina, CH_2Cl_2).
Y. 85%. Viscous oil.
Analytical data were identical with those described earlier.<sup>6</sup>
Preparation of compound 16
Reaction of compound 12 (22 mg, 0.08 mmol) with DMAP (0.8 mg, 0.1 equiv.)
and (BOC)<sub>2</sub>O (28 mg, 1.8 equiv.) using the procedure described for
compound 7 (vide supra) afforded the crude product of compound 16, which
was purified by flash chromatography (alumina, CH_2Cl_2).
Y. 50%. Viscous oil.
IR: 1730 (C=O).
    NMR: 1.65 (9H, s, -C(CH<sub>3</sub>)<sub>3</sub>), 3.68 (3H, s, -CO<sub>2</sub>CH<sub>3</sub>), 7.13-7.40
1<sub>H</sub>
(3H, m, H-8, 9, 10), 8.09 (1H, m, H-11).
MS: 398 (M<sup>+</sup>), 342, 341 (100%), 297, 269, 170, 169; exact mass: 398.2232
(calc. for C<sub>23</sub>H<sub>30</sub>N<sub>2</sub>O<sub>4</sub>: 398.2206).
Preparation of compounds 13 and 15
Hydrogenation (MeOH, PtO2, 17 h) of compound 7 (150 mg, 0.43 mmol)
followed by separation by column chromatography (silica, CH<sub>2</sub>Cl<sub>2</sub>-MeOH,
97:3) afforded compounds 13 and 15.
Compound 13:
Y. 5%. Viscous oil.
Analytical data were identical with those of compound 13 described above.
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Compound 15: Y. 75%. Viscous oil. Analytical data were identical with those of compound 15 described above. Preparation of compounds 14 and 16 Hydrogenation (MeOH, PtO₂, 22 h) of compound 8 (160 mg, 0.40 mmol)followed by separation by column chromatography (silica, CH₂Cl₂-MeOH, 99:1) afforded compounds 14 and 16. Compound 14: Y. 15%. Viscous oil. Analytical data were identical with those of compound 14 described above. Compound 16: Y. 70%. Viscous oil. Analytical data were identical with those of compound 16 described above. Preparation of compounds 9 and 11 by BOC cleavage Compound <u>13</u> or <u>15</u> (88 mg, 0.25 mmol) was dissolved in HCOOH (3.7 ml). The reaction mixture was stirred for 28 h at rt (Ar atm). It was then evaporated and shaken with 10% ag. Na₂CO₂. Extraction with CH₂Cl₂ and drying (Na_2SO_4) afforded the crude product of compound 9 or compound 11, respectively, which was purified by column chromatography (silica, CH₂Cl₂-MeOH, 95:5). Compound 9: Y. 80%. Viscous oil. Analytical data were identical with those described earlier. 6,17,21,22 Compound 11: Mp. $159-161^{\circ}C^{6}$. $160-161^{\circ}C^{22}$. 159-161⁰C (EtOH) Y. 80%. Mp. (lit. 157°C²³). Analytical data were identical with those described earlier.^{6,17,21-23} Preparation of compounds 10 and 12 by BOC cleavage Treatment of compounds 14 and 16 (88 mg, 0.22 mmol) with HCOOH following the procedure described for compounds $\underline{9}$ and $\underline{11}$ (vide supra) gave the crude products of compounds 10 and 12, respectively, which were purified by preparative TLC (silica, CH₂Cl₂-MeOH, 90:10). Compound 10: Y. 80%. Mp. 180-182^OC (MeOH). identical with those described earlier.¹¹ Analytical data were

7626

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Compound 12:
Y. 80%. Mp. 164-165<sup>o</sup>C (MeOH).
Analytical data were identical with those of compound <u>12</u> described above.
Preparation of compound 17
Pyridinium salt <u>1</u> (3.00 g, 9.06 mmol) (vide supra) was dissolved in
MeOH:H_{2O} (9:1, 150 ml), and NaBH<sub>4</sub> (4.13 g, 0.11 mol) was added during
30 min at 0^{\circ}C (Ar atm). The mixture was stirred for 4 h at rt. Usual
work-up afforded compound 17.
Y. 99%. Mp. 122°C (benzene-hexane) (lit.<sup>24</sup> Mp. 122°C).
IR (KBr): 3440 (NH).
<sup>1</sup>H NMR: 1.02 (3H, t, J=7.2 \text{ Hz}, -CH<sub>2</sub>CH<sub>3</sub>), 5.47 (1H, br s, H-4'), 6.92
(1H, d, J=1.8 Hz, H-2), 7.18-7.68 (4H, m, arom. H), 8.54 (1H, br s, NH).
MS: 254 (M<sup>+</sup>), 124 (100%); exact mass: 254.1775 (calc. for C_{17}H_{22}N_2:
254.1783).
Preparation of compound 18
Compound 17 (1.51 g, 5.95 mmol) was reacted with H_2O_2 (30%, 1.7 ml) in
CHCl<sub>3</sub>:MeOH (1:1, 30 ml) to afford
                                             after the usual work-up the
corresponding N-oxide (1.59 g, 99%). The N-oxide in trifluoroacetic acid
(32 ml) was stirred at 0°C (Ar atm) and trifluoroacetic anhydride (8 ml,
9.6 equiv.) was added during 15 min. After 1 h at rt, 2N HCl (48 ml)
                                                                              was
added carefully and the mixture was kept at 70°C for 20 min. Neutrali-
                                   CH2Cl2 followed by usual work-up and
zation with NaHCO<sub>3</sub> (s)
                              in
purification by flash
                              chromatography (alumina,
                                                             CH_2Cl_2)
                                                                        afforded
compound 18.
Y. 20%. Mp. 108-112°C (benzene-hexane) (lit.<sup>25</sup> Mp. 115-116°C).
IR: 3480 (NH).
                (3H, t, J=7.4 Hz, -CH<sub>2</sub>CH<sub>3</sub>), 4.57 (1H, m, H-12b), 5.60
<sup>1</sup>H NMR: 1.13
(1H, m, H-2), 7.00-7.55 (4H, arom. H), 8.17 (1H, br s, NH).
MS: 252 (M<sup>+</sup>, 100%), 251, 237, 223; exact mass: 252.1601 (calc. for
C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>: 252.1626).
Preparation of compound 19
Reaction of compound 18 (34 mg, 0.14 mmol) with DMAP (2 mg, 0.1 equiv.) and
(BOC) 20 (53 mg, 1.8 equiv.) following the procedure described for
compound 7 (vide supra) afforded the crude product of compound 19, which
was purified by column chromatography (alumina, CH<sub>2</sub>Cl<sub>2</sub>).
Y. 90%. Viscous oil.
IR: 1730 (C=O).
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<sup>1</sup>H NMR:
            0.80 (3H, t, J=7.3 Hz, CH<sub>2</sub>CH<sub>3</sub>), 1.66 (9H, s, -CH(CH<sub>3</sub>)<sub>3</sub>),
5.51 (1H, m, H-2), 7.14-7.41 (3H, m, H-8, 9, 10), 8.11 (1H, m, H-11).
MS: 352 (M<sup>+</sup>), 323, 295 (100%), 267, 251; exact mass: 352.2168 (calc. for
C<sub>22</sub>H<sub>28</sub>N<sub>2</sub>O<sub>2</sub>: 352.2151).
Preparation of compound 20
Hydrogenation (MeOH, PtO<sub>2</sub>, 5h) of compound 18 (61 mg, 0.24 mmol) afforded
after usual work-up compound 20.
                                  (benzene-hexane) (lit. Mp. 116-119°C<sup>21</sup>, 61-
Y.
     90%.
             Mp.
                    216-217<sup>o</sup>C
70^{\circ}c^{26}).
Analytical data were identical with those described earlier.<sup>17</sup>
Preparation of compound 21
Reaction of compound 20 (41 mg, 0.16 mmol) with DMAP (2 mg, 0.1 equiv.) and
(BOC)<sub>2</sub>O (64 mg, 1.8 equiv.) following the procedure described for
compound \underline{7} (vide supra) afforded the crude product of compound \underline{21}, which
was purified by flash chromatography (alumina, CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 99.5:0.5).
Y. 80%. Viscous oil.
IR: 1720 (C=O).
1_{\rm H}
      NMR:
              0.75
                       (3H, def, -CH_2CH_3), 1.65 (9H, s, -C(CH_3)_3),
                                                                                   3.74
(1H, m, H-12b), 7.13-7.40 (3H, m, H-8, 9, 10), 8.06 (1H, m, H-11).
     354 (M<sup>+</sup>), 298, 297 (100%), 253, 170, 169; exact mass: 354.2315 (calc.
MS:
for C<sub>22</sub>H<sub>30</sub>N<sub>2</sub>O<sub>2</sub>: 354.2307).
Preparation of compound <u>20</u> by BOC cleavage
Treatment of compound 21 (32 mg, 0.09 mmol) with HCOOH following the
procedure described for compounds 9 and 11 (vide supra) afforded after
normal work-up compound 20.
                    216-217°C (benzene-hexane) (lit. Mp. 116-119°C<sup>21</sup>. 61-
Y.
     95%.
             Mp.
70^{\circ}C^{26}).
Analytical data were identical with those of compound 20 described above.
Preparation of compound 22
Hydrogenation (MeOH, PtO<sub>2</sub>, 60 h) of compound 19 (83 mg, 0.24 mmol)
followed by purification by flash chromatography (alumina, CH<sub>2</sub>Cl<sub>2</sub>)
afforded compound 22.
Y. 35%. Viscous oil.
IR: 1730 (C=O).
<sup>1</sup>H NMR: 0.90 (3H, def, -CH_2CH_3), 1.66 (9H, s, -C(CH_3)_3) 4.64 (1H,
m, H-12b), 7.11 - 7.41 (3H, m, H-8, 9, 10), 8.07 (1H, m, H-11).
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MS: 354 (M⁺), 298, 297 (100%), 253, 170, 169; exact mass: 354.2346 (calc. for $C_{22}H_{30}N_2O_2$: 354.2307).

Preparation of compound 23 by BOC cleavage Treatment of compound 22 (32 mg, 0.09 mmol) with HCOOH following the procedure described for compounds 9 and 11 (vide supra) gave after normal work-up compound 23. Y. 95%. Mp. 123-125°C (benzene-hexane) (lit. Mp. $90-93°C^{21}$, 113-114°C²⁶). Analytical data were identical with those described earlier.¹⁷

Preparation of compound $\underline{22}$ from compound $\underline{23}$ Reaction of compound $\underline{23}$ (23 mg, 0.09 mmol) with DMAP (1.1 mg, 0.1 equiv.) and (BOC)₂O (35 mg, 1.8 equiv.) following the procedure described for compound 7 (<u>vide supra</u>) afforded the crude product of compound $\underline{22}$, which was purified by flash chromatography (alumina, CH₂Cl₂). Y. 45%. Viscous oil. Analytical data were identical with those of compound $\underline{22}$ described above.

Preparation of compound <u>11</u> from (±)-deplancheine <u>24</u> Hydrogenation (MeOH, PtO₂, 20 h) of (±)-deplancheine <u>24</u>¹⁶ (21 mg, 0.08 mmol) gave after normal work-up compound <u>11</u>. Y. 99%. Mp. 159-161°C (EtOH) (lit. Mp. 159-161°C⁶, 160-161°C²², 157°C²³). Analytical data were identical with those described earlier.^{17,21-23}

Preparation of compound $\frac{25}{24}$ Reaction of (±)-deplancheine $\underline{24}^{16}$ (28 mg, 0.11 mmol) with DMAP (1.4 mg, 0.1 equiv.) and (BOC)₂O (44 mg, 1.8 equiv.) following the procedure described for compound 7 gave compound 25. The crude product was purified by flash chromatography (alumina, CH₂Cl₂). Y. 75%. Viscous oil. IR: 1735 (C=O). ¹H NMR: 1.68 (9H, s, -C(CH₃)₃), 4.41 (1H m, H-12b), 5.34 (1H, m, =<u>CH</u>CH₃), 7.13-7.50 (3H, m, H-8,9,10), 8.11 (1H, m, H-11). MS: 352 (M⁺), 296, 295 (100%), 281, 251, 170, 169; exact mass: 352.2162 (calc. for C₂₂H₂₈N₂O₂: 352.2151).

Preparation of compounds $\underline{13}$ and $\underline{15}$ from compound $\underline{25}$ Hydrogenation (MeOH, PtO₂, 20 h) of compound $\underline{25}$ (29 mg, 0.08 mmol)

followed by purification by flash chromatography (alumina, CH_2Cl_2) afforded a 1:1 mixture of compounds 13 and 15. Y. 50%. For the analytical data, see above. REFERENCES AND NOTES Lounasmaa, M.; Jokela, R. Tetrahedron 1989, 45, 3975. 1. Hoshino, T.; Shimodaira, K. <u>Liebigs Ann. Chem.</u> 1935, <u>520</u>, 19. Fry, E.M. <u>J. Org. Chem.</u> 1964, <u>29</u>, 1647. 2. 3. Fry, E.M.; Beisler, J.A. J. Org. Chem. 1970, 35, 2809. 4. Lounasmaa, M. in "Studies in Natural Products Chemistry", ed. Atta-ur-Rahman, Vol. 1, Stereoselective Synthesis (Part A), Elsevier, Amsterdam, 1988, pp. 89-122. 5. Tamminen, T.; Jokela, R; Tirkkonen, B; Lounasmaa, M. Tetrahedron 6. 1989, <u>45</u>, 2683. Grierson, D.S.; Harris, M.; Husson, H.-P. J. Am. Chem. Soc. 1980, 7. 102, 1064. Jokela, R.; Tamminen T.; Lounasmaa, M. <u>Heterocycles</u> 1985, <u>23</u>, 1707. 8. Grierson, D.S.; Harris, M.; Husson, H.-P. Tetrahedron 1983, 39, 9. 3683. Lounasmaa, M.; Karvinen, E.; Koskinen, A.; Jokela, R. 1987, <u>43</u>, 2135. 10. Tetrahedron Jokela, R.; Karvinen, E.; Tolvanen, A.; Lounasmaa, M. Tetrahedron 11. 1988, <u>44</u>, 2367. 12. Lounasmaa, M.; Jokela, R. <u>Heterocycles</u> 1986, <u>24</u>, 1663. Cavé, A.; Kan-Fan, C.; Potier, P.; Le Men, J. Tetrahedron 1967, 23, 13. 4681. Potier, P. <u>Rev. Latinoamer. Quim.</u> 1978, 9, 47. Lounasmaa, M.; Koskinen, A. <u>Heterocycles</u> 1984, <u>22</u>, 1591. Hämeilä, M.; Lounasmaa, M. <u>Acta Chem. Scand.</u> 1981, <u>B35</u>, 217. Lounasmaa, M.; Jokela, R.; Tamminen, T. <u>Heterocycles</u> 1985, <u>23</u>, 1367. See also, Ref. 6, Note 15 and Ref. 11, Note 15. 14. 15. 16. 17. 18. Thal, C; Imbert, T.; Husson, H.-P.; Potier, P. Bull. Soc. Chim. France 1973, 2010. 19. Jokela, R.; Juntunen, A.; Lounasmaa, M. Planta Medica 1987, 386. 20. Wenkert, E.; Massy-Westropp, R.A.; Lewis, R.G. J. Am. Chem. Soc. 1962, <u>84</u>, 3732 Wenkert, E.; Wickberg, B. <u>J. Am. Chem. Soc.</u> 1962, <u>84</u>, 4914. 21. 22. Yamanaka, E.; Narushima, M.; Inukai, K.; Sakai, S.-I. Chem. Pharm. <u>Bull.</u> 1986, <u>34</u>, 77. 23. Massiot, G.; Oliveira, F.S.; Lévy, J. Bull. Soc. Chim. France_II 1982, 185. 24. Grierson, D.S; Vuilhorgne, M.; Husson, H.-P. J. Org. Chem. 1982, <u>47</u>, 4439. Chevolot, L.; Husson, A.; Kan-Fan, C.; Husson, H.-P.; Potier, P. 25. Bull. Soc. Chim. France 1976, 1222. Laronze, J.Y.; Laronze, J.; Caron, B.; Lévy, J.; Le Men, J. Bull. 26.

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